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Hermann Sicius

Edelgase

Eine Reise durch das Periodensystem



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Edelgase Hermann Sicius, 2015-06-05 Hermann Sicius bringt dem Leser in knapper Form alle wichtigen Informationen rund um die Edelgase nahe. Die sechs bekannten Mitglieder dieser achten Hauptgruppe von Helium bis Radon sind bei Raumtemperatur allesamt gasförmig, nur bei den niedrigen Aggregatzustand des künstlich erzeugten Ununoctiums wird noch diskutiert. Das bekannteste Einsatzgebiet für alle Edelgase mit Ausnahme des Radons ist das als Füllgas in

Gasentladungslampen und speziell des Heliums als Füllgas für Ballons. Aber was gibt es darüber hinaus? Welche chemischen Verbindungen der Edelgase existieren bereits? Der Autor beschreibt die gegenwärtige Lage und gibt einen Ausblick in die Zukunft.

Chromgruppe: Elemente der sechsten Nebengruppe Hermann Sicius, 2016-06-23 Hermann Sicius stellt in diesem essential die Elemente der sechsten Nebengruppe Chromgruppe ausführlich vor. Er zeigt, dass die Elemente dieser Gruppe chemisch meist relativ reaktionsträge sind und regelmäßige Abstufungen ihrer Eigenschaften zeigen. So nehmen vom Chrom zum Wolfram Dichte, Schmelz- und Siedepunkte zu, die Reaktivität leicht ab. Chrom, Molybdän und Wolfram entdeckte man Ende des 18. Jahrhunderts, das nur künstlich darstellbare Seaborgium erst etwa 200 Jahre später. Molybdän ist z.B. Bestandteil sehr harter Stähle, Wolfram wurde für Wendeln von Glühlampen verwendet.

Vanadiumgruppe: Elemente der fünften Nebengruppe Hermann Sicius, 2016-06-14 Hermann Sicius stellt in diesem essential die Elemente der fünften Nebengruppe Vanadiumgruppe ausführlich vor. Er zeigt, dass diese Elemente teilweise reaktiv, aber gelegentlich auch sehr reaktionsträge sind, dass sie an der Luft lagernd von einer dicken passivierenden Oxidschicht vor weiterer Korrosion geschützt und auch in Säuren nur selten und dann nur unter Anwendung drastischer Methoden löslich sind. Die technische Anwendung zeigt der Autor für diese Elemente auf: die alle sehr harte, hochschmelzende Metalle sind und deren Verbindungen mit Kohlen und Stickstoff teils extrem hochschmelzende harte Schleifmaterialien darstellen, die man auch industriell einsetzt. Vanadium findet sich in Stahllegierungen, Tantal in medizinischen Implantaten, Ferroniob als Bestandteil von Gasturbinen oder Vanadiumdioxid als Katalysator bei der Herstellung von Schwefelsäure.

Zinkgruppe: Elemente der zweiten Nebengruppe Hermann Sicius, 2017-04-24 Hermann Sicius stellt ausführlich die Elemente der zweiten Nebengruppe vor: Zink, Cadmium, Quecksilber, Copernicium, deren physikalische und chemische Eigenschaften relativ ähnlich sind. Auswirkungen der Lanthanoidenkontraktion erkennt man hier kaum; noch Cadmium steht in seinen Eigenschaften etwa zwischen dem Zink und dem Quecksilber. Zink und Cadmium haben negative Normalpotentiale, wogegen Quecksilber ein Halbedelmetall ist. Die Elemente dieser Gruppe geben meist ein oder zwei äußere Valenzelektronen ab, um eine stabile Elektronenkonfiguration zu erreichen. Für das höchste Element dieser Nebengruppe, das Copernicium, konnten bisher kaum chemische Untersuchungen durchgeführt werden. Es ist zu erwarten, dass es sich chemisch ähnlich wie Quecksilber verhält. Zink als Element kennt man seit dem 17. Jahrhundert, Cadmium seit 1817, wogegen Quecksilber schon in der Antike bekannt war. Die erstmalige Darstellung von Atomen des Coperniciums gelang 1996.

Edelgase Hermann Sicius, 2022

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