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Rachel Sandford



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 solução $\tau = 140 \text{ dias}$
 $\lambda = \frac{\ln 2}{\tau} = \frac{\ln 2}{140} \approx 0,00495 \text{ dias}^{-1}$
 $\ln \frac{A}{A_0} = -\lambda t$
 $\ln \frac{1}{4} = -\lambda t$
 $\ln 1 - \ln 4 = -\lambda t$
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 $t = \frac{1,3863}{\lambda} = \frac{1,3863}{0,00495} \approx 280 \text{ dias}$

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 un exceso de partículas emite partículas y o radiación hasta alcanzar una configuración estable el exceso de partículas se
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 partícula alfa emissão de partícula beta emissão de pósitron captura de elétron e fissão espontânea cada um desses tipos de
 decaimento está relacionado com as características da estrutura nuclear de um dado radionuclídeo

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 $\ln 2 = \lambda \tau$
 $\lambda = \frac{\ln 2}{\tau}$
 $\lambda = \frac{\ln 2}{0,693} \approx 1 \text{ ano}^{-1}$
 $n = \frac{A}{\lambda} = \frac{12}{1} = 12$
 o número n de núcleos de ^{12}C em 1

g de carbono é 6 02 1023 núcleos mol 12 g mol n 1 g n 6 02 1023 12 5 02 1022 núcleos g o número de núcleos de 14c radioativo é então igual a razão 1 3 10 12

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