

4. a. A sample of a $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$ solid salt mixture is dissolved in ~150 mL of deionized water previously adjusted to a pH close to 8.0. The precipitate, after having been filtered, was air-dried and weighed. Data for Trial 1 were obtained as shown. Complete the following table. (Your Report Sheet's Record column values match the constant number of significant figures.)

- A. Precipitation of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ from the Salt Mixture
- Mass of salt mixture (g)
 - Mass of filter paper (g)
 - Mass of filter paper and $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)
 - Mass of air-dried $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)

Calculation Table

Data Analysis, 1.

0.879
1.886
2.180
1.301

B. Determination of Limiting Reactant

- Limiting reactant in salt mixture
- Excess reactant in salt mixture

$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$
 $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$

Data Analysis, 2.

Data Analysis

- Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ precipitated (mol)
Moles calculation: $1 \text{ mol} / 146.13 =$
- Moles of limiting reactant in salt mixture (mol)
See equation 3.1: $1 \text{ mol} / 1 \text{ mol} \text{ CaCl}_2$
- Mass of limiting reactant in salt mixture (g)
Moles calculation: $147.02 \text{ g} / 1 \text{ mole} = 287.56 \text{ g CaCl}_2$
- Mass of excess reactant in salt mixture (g)
Equals mass of salt mixture minus mass of limiting reactant.
Moles calculation: $2.879 - 0.664 = 2.215$
- Percent limiting reactant in salt mixture (%)
Moles calculation: $0.286 / 0.879 = 32.54$
- Percent excess reactant in salt mixture (%)

0.00945 mole
0.00945 mole
CaC₂O₄ · H₂O
CaCl₂

Data Analysis, 3.

0.595 g $\text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$
32.5%
67.5%

4. b. For Trials 2 and 3, the percent $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the mixture was 39.7% and 28.4% respectively.
- What is the average percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? Data Analysis, B.

- What are the standard deviation and the relative standard deviation (%RSD) for the percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? Data Analysis, C and D.

Exp 8 Limiting Reactant Prelab Answers

**John R. Holum, Ruth C. Denison, Sandra
L. Olmsted**



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