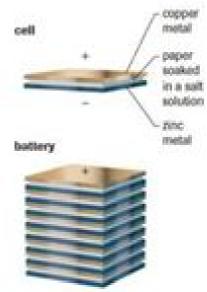
# Introduction to Electrochemistry

- An electric cell converts chemical energy into electrical energy
  - Alessandro Volta invented the first electric cell but got his inspiration from Luigi Galvani. Galvani's crucial observation was that two different metals could make the muscles of a frog's legs twitch. Unfortunately, Galvani thought this was due to some mysterious "animal electricity". It was Volta who recognized this experiment's potential.
  - An electric cell produces very little electricity, so Volta came up with a better design:
- A battery is defined as two or more electric cells connected in series to produce a steady flow of current
  - Volta's first battery consisted of several bowls of brine (NaCl<sub>(aq)</sub>) connected by metals that dipped from one bowl to another
  - His revised design, consisted of a sandwich of two metals separated by paper soaked in salt water.





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John Michels (Journalist)

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