



Reaction Energy



Tools: REACTION INVESTIGATION

Starting temperature: 21.0 °C

CONTROL

Reaction 1

$$2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$$

☒ Forward ☐ Reverse

Key

H ₂		H ₂ O	
O ₂			

Instructions

Click **Play** to start the reaction. The reaction chamber contains 0.2 moles of H₂ and 0.1 moles of O₂.

Theoretical ΔH : — kJ

Experimental ΔH : — kJ



ASSESSMENT QUESTIONS

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Questions & Answers

1. A reaction occurs in a calorimeter, resulting in the starting and final temperatures shown below. What can you say about the reaction and the enthalpy change (ΔH) during the reaction?

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Peipei Pang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Gizmo Explore Learning Guide Answers Phase Change:

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