

## Chemistry EOC Review – Answer Key

- 1) a. 68,200,000 cg      b. 5000 mm      c. 0.0125 kL      d. 0.548 dm
- 2) ~21.5 mL
- 3) production of heat & light, formation of a gas, formation of a precipitate, color change
- 4)  $33.64 \text{ g} / 4.28 \text{ ml} = 7.86 \text{ g/ml} = \text{Fe}$
- 5) m.p.  $-183^{\circ}\text{C} = \text{CH}_4$  (methane)
- 6)  $\text{H}_2$  – splint test (burning splint ignites hydrogen)  
 $\text{O}_2$  – splint test (oxygen reignites glowing splint)  
 $\text{H}_2\text{O}$  – cobalt chloride paper (paper changes from blue to pink in the presence of water vapor)  
 $\text{CO}_2$  – splint test (carbon dioxide extinguishes burning splint);  
limewater test (limewater turns cloudy when carbon dioxide is bubbled through it,  
producing  $\text{CaCO}_3$ )
- 7) The protons are positively charged particles found inside the nucleus along with the neutrons, which have no charge. The total number of these nucleons in an atom is the mass # / atomic mass of the atom. The electrons are the negatively charged particles found orbiting in "clouds" around the nucleus. Electrons can be gained by the atom to form anions or lost to form cations.
- 8) The thing that makes any element on the periodic table different from all other elements is the number of protons in its nucleus. Isotopes are atoms of the same type of element which have different masses because they have different numbers of neutrons.
- 9) a.  $^{39}_{19}\text{K}^+$   
b.  $^{57}_{26}\text{Fe}$   
c.  $^{33}_{16}\text{S}^{2-}$
- 10)
- | <u>chemical symbol</u> | <u>atomic number</u> | <u>mass number</u> | <u># of protons</u> | <u># of electrons</u> | <u># of neutrons</u> | <u>net charge</u> |
|------------------------|----------------------|--------------------|---------------------|-----------------------|----------------------|-------------------|
| $\text{K}^+$           | 19                   | 42                 | 19                  | 18                    | 23                   | +1                |
| Sr                     | 38                   | 88                 | 38                  | 38                    | 50                   | 0                 |
| $\text{F}^-$           | 9                    | 20                 | 9                   | 10                    | 11                   | -1                |
| P                      | 15                   | 31                 | 15                  | 15                    | 16                   | 0                 |
| $\text{Ca}^{2+}$       | 20                   | 42                 | 20                  | 18                    | 22                   | +2                |
- 11) uranium – 238       $238 - 92 = 146$  neutrons
- 12) (1) 199.59g  
(2) They contain the same # of protons (they are the same element), but different #s of neutrons (electrons don't matter, b/c they have virtually no mass), & thus different masses.

# Eoc Chemistry Practice Key

**Thirumalaisamy P.**  
**Velavancorresponding**



## Eoc Chemistry Practice Key:

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**Laboratory** ,2002 **The United States Army and Navy Journal and Gazette of the Regular and Volunteer Forces** ,1866 *Books in Print Supplement* ,2002 *A European Strategy for Jobs and Growth* Great Britain: Parliament: House of Lords: European Union Committee,2006-03-16 This report by the European Union Committee examines the progress of the European Union in initiating a strategy for jobs and growth across the Community as a whole The background to this report stems from the Spring European Council meeting in 2000 in Lisbon and the launch of an economic reform agenda The Committee observes that since the Lisbon Agenda little progress has been made and the performance of many of the larger European economies has been poor The Committee has noted that certain worrying signs of protectionist behaviour have developed especially regarding barriers of cross border mergers The EU has recognized this weak performance and the Agenda was relaunched in 2005 with a greater focus on the key economic priorities of more growth and jobs Also all Member States are now required to produce an annual National Action Plan highlighting the policies being pursued to improve economic growth and increase employment The Committee sets out a number of recommendations to further push forward the priorities of growth and jobs including that the Commission should seek to complete the progress towards an internal market that Member States should influence one another in the development of good practice through statistical comparison of their economic progress and agree on quantifiable targets that the format of the National Action Plans should include not only the successes but where countries are underperforming that the Agenda be given a higher public profile **Corrosion**

,1992 Issues include special section called Corrosion abstracts      *Texas EOC Chemistry - the All-In-one Guide* Mary Amboji,2013-03-19 The complete guide for Texas EOC Chemistry has hundreds of guided and independent practice questions based on the categories tested on the EOC The dual coded questions help students gain mastery over scientific process skills Five category based tests are used to further hone your skills Calculator screen shots help you learn how to maximize your efficiency to solve complex problems There are two full length tests to prepare you for the actual exam This book can be used to prep students on chemistry      **South Dakota Chemistry Eoc Success Strategies Study Guide** South Dakota Eoc Exam Secrets Test Prep,2014-08-22 South Dakota Chemistry EOC Success Strategies helps you ace the South Dakota End of Course Exams without weeks and months of endless studying Our comprehensive South Dakota Chemistry EOC Success Strategies study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you ve ever imagined South Dakota Chemistry EOC Success Strategies includes The 5 Secret Keys to South Dakota EOC Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don t Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don t Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific South Dakota EOC exam and much more      **STAAR EOC Chemistry Assessment Secrets Study Guide** Mometrix Media LLC,2016-10-24 Includes Practice Test Questions STAAR EOC Chemistry Assessment Secrets helps you ace the State of Texas Assessments of Academic Readiness without weeks and months of endless studying Our comprehensive STAAR EOC Chemistry Assessment Secrets study guide is written by our exam experts who painstakingly researched every topic and concept that you need to know to ace your test Our original research reveals specific weaknesses that you can exploit to increase your exam score more than you ve ever imagined STAAR EOC Chemistry Assessment Secrets includes The 5 Secret Keys to STAAR Success Time is Your Greatest Enemy Guessing is Not Guesswork Practice Smarter Not Harder Prepare Don t Procrastinate Test Yourself A comprehensive General Strategy review including Make Predictions Answer the Question Benchmark Valid Information Avoid Fact Traps Milk the Question The Trap of Familiarity Eliminate Answers Tough Questions Brainstorm Read Carefully Face Value Prefixes Hedge Phrases Switchback Words New Information Time Management Contextual Clues Don t Panic Pace Yourself Answer Selection Check Your Work Beware of Directly Quoted Answers Slang Extreme Statements Answer Choice Families Along with a complete in depth study guide for your specific STAAR exam and much more      **Who's who Among African Americans** Ashyia N. Henderson,2000 Devoted to recording

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