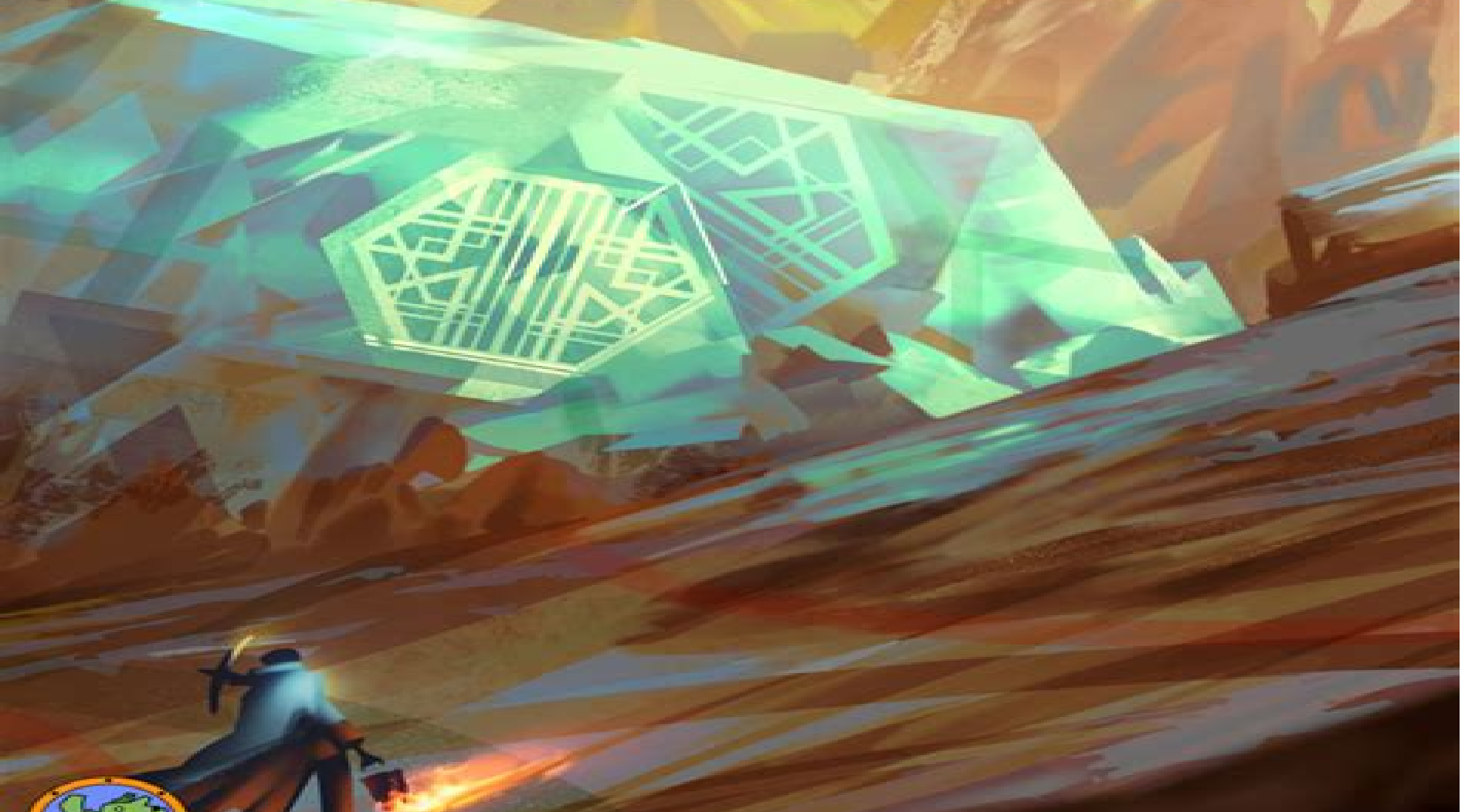


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Manual Fantasy Series

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 $80\text{ g I}_2\text{O}_5 \times \frac{1\text{ mol I}_2\text{O}_5}{1\text{ mol I}_2\text{O}_5} \times \frac{1\text{ mol I}_2}{1\text{ mol I}_2\text{O}_5} \times \frac{333.8\text{ g I}_2}{1\text{ mol I}_2} = 333.8\text{ g I}_2$
 $28\text{ g CO} \times \frac{1\text{ mol CO}}{28\text{ g CO}} \times \frac{1\text{ mol I}_2}{5\text{ mol CO}} = 0.2\text{ mol I}_2 \times 333.8\text{ g I}_2/\text{mol I}_2 = 66.8\text{ g I}_2$

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