

# Laboratory Exercise: Double Replacement Reactions...

TABLE II: SCALE UP OF PRECIPITATION REACTIONS

Row #	Chemical Formula of Reagent	Chemical Formula of Reagent	List the four ions present in reagents		insoluble	soluble
					Formula of Solid Product appearance in test tube	Formula of Aqueous Product appearance in test tube
2	NaOH	FeCl <sub>3</sub>	Na <sup>+</sup>	OH <sup>-</sup>	Fe(OH) <sub>3</sub>	
			Fe <sup>3+</sup>	Cl <sup>-</sup>	very small orange	very cloudy orange
3	NaNO <sub>3</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	NO <sub>3</sub> <sup>-</sup>		
			Fe <sup>3+</sup>	Cl <sup>-</sup>	solid white powder	yellow cloudy
5	Na <sub>2</sub> CO <sub>3</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	CO <sub>3</sub> <sup>2-</sup>		
			Fe <sup>3+</sup>	Cl <sup>-</sup>	solid orange powdery	very cloudy orange
6	Na <sub>3</sub> PO <sub>4</sub>	FeCl <sub>3</sub>	Na <sup>+</sup>	PO <sub>4</sub> <sup>3-</sup>		
			Fe <sup>3+</sup>	Cl <sup>-</sup>	yellowish small	yellow cloudy
12	AgNO <sub>3</sub>	NaOH				
					not almost brown	cloudy brown
13	NaOH	CuSO <sub>4</sub>	N			
					gel like chunk	blue uncent
17	AgNO <sub>3</sub>	Na <sub>2</sub> CO <sub>3</sub>				
					white flake small	white cloudy
18	AgNO <sub>3</sub>	Na <sub>3</sub> PO <sub>4</sub>				
					yellow egg	clear yellow
19	CuSO <sub>4</sub>	Na <sub>2</sub> CO <sub>3</sub>				
					small blue/white powder	cloudy blue
20	CuSO <sub>4</sub>	Na <sub>3</sub> PO <sub>4</sub>				
					very small blue particles	cloudy light-blue

# Experiment 12 Double Replacement Reactions With Answers

**Katherine J. Denniston, Denniston  
Katherine**



## **Experiment 12 Double Replacement Reactions With Answers:**

CHEMISTRY EXPERIMENTS James Signorelli, 2014-09-19 Gifted and talented students and any student interested in pursuing a science major in college needs a rigorous program to prepare them while they are still in high school This book utilizes a format where the application of several disciplines science math and language arts principles are mandated Each lab concludes with either an essay or a detailed analysis of what happened and why it happened This format is based on the expectations of joining a university program or becoming an industrial science professional The ideal student lab report would be written in a lab research notebook and then the essay or final analysis is done on a word processor to allow for repeat editing and corrections The research notebook has all graph pages a title section and a place for the students and their assistants to sign and witness that exercise The basic mechanics of the lab report title purpose procedure diagrams data table math and calculations observations and graphs are handwritten into the book The conclusion is done on a word processor MS Word which allows the instructor to guide the student in writing and editing a complete essay using the MLA format When the final copy is completed the essay is printed and inserted into the lab notebook for grading At the end of the term the student has all their labs in one place for future reference These lab notebooks can be obtained for as little as 3 00 per book This is money well spent In our district the Board of Education buys the books for each student The BOE sees these books as expendable but necessary materials for all science and engineering instruction **E3 Chemistry Guided Study**

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Martha Mackin,2012-12-02 Study Guide to Accompany

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*Study Guide with Selected Solutions* Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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