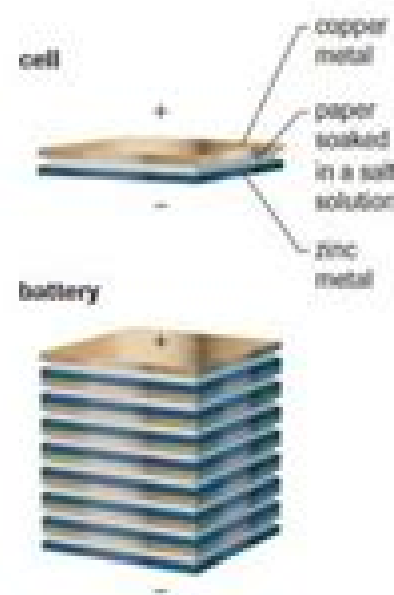


Introduction to Electrochemistry

- ▶ An *electric cell* converts chemical energy into electrical energy
 - ▶ Alessandro Volta invented the first electric cell but got his inspiration from Luigi Galvani. Galvani's crucial observation was that two different metals could make the muscles of a frog's legs twitch. Unfortunately, Galvani thought this was due to some mysterious "animal electricity". It was Volta who recognized this experiment's potential.
 - ▶ An electric cell produces very little electricity, so Volta came up with a better design:
- ▶ A *battery* is defined as two or more electric cells connected in series to produce a steady flow of current
 - ▶ Volta's first battery consisted of several bowls of brine ($\text{NaCl}_{(\text{aq})}$) connected by metals that dipped from one bowl to another
 - ▶ His revised design, consisted of a sandwich of two metals separated by paper soaked in salt water.



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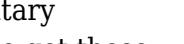
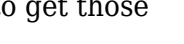

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