Name: Date: 3/3/2020

Student Exploration: Ionic Bonds

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Answer: Nate can give Clara one marker so they can both have 8 markers.

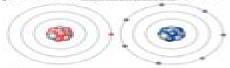
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Answer: Maggie can give one to both Fred and Florence.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lonic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (CI) are selected from the menus at right. Click Play (L) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



 Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons are called valence electrons. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

- - A. Try pulling an electron away from each atom. Based on this experiment, which atom

is a metal?

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John Bonner, George William
Curtis, Henry Mills Alden, Samuel
Stillman Conant, Montgomery
Schuyler, John Foord, Richard Harding
Davis, Carl Schurz, Henry Loomis
Nelson, John Kendrick Bangs, George
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Prentice Hall Chemistry ,2000 Resources for Teaching Middle School Science Smithsonian Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents The Origins of Evolutionary Innovations Andreas Wagner, 2011-07-14 The history of life is a nearly four billion year old story of transformative change This change ranges from dramatic macroscopic innovations such as the evolution of wings or eyes to a myriad of molecular changes that form the basis of macroscopic innovations. We are familiar with many examples of innovations qualitatively new phenotypes that provide a critical benefit but have no systematic understanding of the principles that allow organisms to innovate This

book proposes several such principles as the basis of a theory of innovation integrating recent knowledge about complex molecular phenotypes with more traditional Darwinian thinking Central to the book are genotype networks vast sets of connected genotypes that exist in metabolism and regulatory circuitry as well as in protein and RNA molecules The theory can successfully unify innovations that occur at different levels of organization It captures known features of biological innovation including the fact that many innovations occur multiple times independently and that they combine existing parts of a system to new purposes It also argues that environmental change is important to create biological systems that are both complex and robust and shows how such robustness can facilitate innovation Beyond that the theory can reconcile neutralism and selectionism as well as explain the role of phenotypic plasticity gene duplication recombination and cryptic variation in innovation Finally its principles can be applied to technological innovation and thus open to human engineering endeavours the powerful principles that have allowed life s spectacular success SAT Subject Test: Chemistry with Online Tests Joseph A. Mascetta, Mark Kernion, 2018-09-01 Always study with the most up to date prep Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product SAT Subject Test Chemistry Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron's SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory ONLINE PRACTICE TESTS Students who purchase this book or package will also get access to two additional full length online SAT Chemistry subject tests with all questions answered and explained **Prentice Hall Science Explorer** Michael J. Padilla.2002 Introduction to Modern Inorganic Chemistry Kenneth Malcolm Mackay, Rosemary Ann Mackay, 1989

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