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**Philip Lee Phillips**



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regional names and international groupings are not included since these can be found conveniently elsewhere The five hundred and seventy two additional entries to this dictionary together with a few minor modifications to the existing material are the result of extensive correspondence and discussion since the appearance of the first **International Geographical History of the World, Accompanying the "International Flat Globe"** ... Robert Edwin Peary, 1905 Course of Study in Geography Cincinnati Public Schools, 1925 *The Position of Geography in British Universities* Sir John Scott Keltie, 1921 Bulletin of the American Geographical Society American Geographical Society of New York, 1915 The Journal of the Manchester Geographical Society, 1886 *Field, Laboratory, and Library Manual in Physical Geography* Cary Thomas Wright, 1906 **Complete Geography** Harmon Bay Niver, 1922 *Geography* Baltimore (Md.). Department of Education, 1924 A Dictionary of Geography, Second Edition F. J. Monkhouse, 2008 The geographer seeks to describe the diverse features of the earth's surface to explain if possible how these features have come to be what they are and to discuss how they influence the distribution of man with his diverse activities Geography therefore stands transitionally yet centrally between the natural sciences the social studies and the humanities While in its concept and content it is an integrated whole of necessity it impinges on the associated disciplines and inevitably makes use of a wide range of kindred terminology In compiling the 3 400 entries for this dictionary the main criterion for inclusion has been usage Geographical textbooks and periodicals have been systematically combed and where a term has been used in a specific geographical context or in a specialist sense which differs from general practice or popular usage it has been included Foreign words are listed where they have been accepted into English geographical literature especially where no satisfactory translation exists Cross references are freely given printed in small capitals where it is necessary to assist the user in tracing cognate and supplementary entries or where the meaning of the word thus shown is essential to the understanding of the entry The emphasis throughout is on specific factual information conveniently accessible on a strict alphabetical basis rather than a bare definition Statistical material and formulae are appended where it would seem helpful in the form of tables under the relevant entries Since this dictionary is neither a gazetteer nor a compendium of current affairs lists of countries and capitals regional names and international groupings are not included since these can be found conveniently elsewhere The five hundred and seventy two additional entries to this dictionary together with a few minor modifications to the existing material are the result of extensive correspondence and discussion since the appearance of the first edition This edition originally published in 1970 has continuing relevance as a resource for professionals F J Monkhouse was formerly chair of geography at the University of Southampton Throughout his career he was also visiting professor at the University of Illinois University of Maryland Queen's University and the Miami University Ohio His books include *The Countries of North Western Europe* and *The Lake District* *The Geographical Teacher*, 1902 **Proceedings of the Royal Geographical Society and Monthly Record of Geography** Royal Geographical Society (Great Britain), 1889 *Proceedings of the Royal Geographical*

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this video accompanies the chm 252 lab nitration of bromobenzene in this video students will get an introduction to the lab as well as the set up for conducting the reaction students should

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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nitration of bromobenzene lab report improved essays 731 words 3 pages open document essay sample check writing quality show more this experiment was performed in order to demonstrate the principles of nitration through electrophilic aromatic substitution it also examined the effects of bromine on an aromatic ring ortho para vs meso directing

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 o

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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lab report 15 nitration of bromobenzene brynn abbott amp katie muerer bea0006 uah ch 336 02 september 21 2022 abstract in this experiment the electrophilic aromatic substitution mechanism was used to obtain 4 nitrobromobenzene and 2 nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no<sub>2</sub><sup>+</sup> which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0 ml of nitric acid and 4.0 ml of sulfuric acid and will then react with 4.5 ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

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lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

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chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially

produced chemicals by adding different functional groups into an aromatic ring

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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