

(Solution Manual all Chapters)

SOLUTIONS CHAPTER 1**1.1. Show that Equation (1.6) follows from Equation (1.3).**Solution: Equation (1.3) is $dE_p = \frac{q^2}{4\pi\epsilon_0 r^2} dr$. Integrating both sides we obtain

$$\int dE_p = \int \frac{q^2}{4\pi\epsilon_0 r^2} dr = \frac{q^2}{4\pi\epsilon_0} \int \frac{1}{r^2} dr = \frac{q^2}{4\pi\epsilon_0} \left[-\frac{1}{r} \right] + \text{const} = E_p$$

To find the constant, we employ the boundary condition that at $r = \infty$, $E_p = E_{\infty}$:

$$E_p(r = \infty) = E_{\infty} = \frac{-q^2}{4\pi\epsilon_0 \cdot \infty} + \text{const} = 0 + \text{const}$$

$$\therefore \text{const} = E_{\infty} \text{ and}$$

$$E_p = E_{\infty} - \frac{q^2}{4\pi\epsilon_0 \cdot r} = E_{\infty}, \text{ Equation (1.6)}$$

1.2. Consider a lithium nucleus, of charge $+3q$. Calculate the first three electron energies for an electron in a Li^{++} ion, using the Bohr model.We repeat the analysis that we used for the hydrogen atom, except that now the charge of the nucleus Q_1 is equal to $3q = +3(1.6 \times 10^{-19})C$. The results of the key steps are

$$F = \frac{Q_1 Q_2}{4\pi\epsilon_0 r^2} = \frac{-3q^2}{4\pi\epsilon_0 r^2}$$

$$E_p(r) = E_{\infty} - \frac{3q^2}{4\pi\epsilon_0 r}$$

$$\frac{mv^2}{r} - \frac{3q^2}{4\pi\epsilon_0 r^2} = 0$$

$$mv_s r_s = n\hbar$$

$$v_s = \frac{3q^2}{(4\pi\epsilon_0)\hbar} \left(\frac{1}{n} \right)$$

$$r_s = \frac{(4\pi\epsilon_0)\hbar^2}{3mq^2} (n^2)$$

$$E_s = \frac{m3^2 q^4}{2(4\pi\epsilon_0)^2 \hbar^2} \left(\frac{1}{n^2} \right)$$

$$E_s = E_{ps} + E_{ks} = E_{\infty} - \frac{9mq^4}{2(4\pi\epsilon_0)^2 \hbar^2} \left(\frac{1}{n^2} \right)$$

Thus $E_1 = E_{\infty} - 9(13.6eV) = E_{\infty} - 122eV$

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