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web at standard temperature and pressure stp the molar volume  $V_m$  is the volume occupied by one mole of a chemical element or a chemical compound it can be calculated by dividing the molar mass  $M$  by mass density  $\rho$  molar gas volume is one mole of any gas at a specific temperature and pressure has a fixed volume molar volume formula

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web this volume is known as the molar gas volume  $V_m$  the volume of a gas may be calculated from its number of moles using volume of gas moles  $V_m$  rearranging this moles of gas  $\frac{V}{V_m}$

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web jun 8 2023  $V = 1.00 \text{ mol} \times 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1} \times 273 \text{ K} = 22.4 \text{ L}$  in other words 1 mole of a gas will occupy 22.4 L at stp assuming ideal gas behavior at stp the volume of a gas is only dependent on number of moles of

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