Name	

Date _____

Pre-laboratory Assignment

- 1. How does temperature affect the kinetic energy of gas molecules?
- 2. What are the names and equations for the three gas laws tested in this lab?

- 3. Complete the following statements:
 - a. When T is constant and volume increases, the pressure _____
 - b. When T is constant and volume decreases, the pressure ____
 - c. When V is constant and temperature increases, the pressure ____
 - d. When V is constant and temperature decreases, the pressure
 - e. When P is constant and temperature increases, the volume _
 - f. When P is constant and temperature decreases, the volume
 - 4. A sample of helium has a volume of 325 mL and a pressure of 655 mm Hg. What will be the pressure if the helium is compressed to 125 mL (T constant)?

5. A gas with a volume of 525 mL at a temperature of -25°C is heated to 75°C. What is the new volume of the gas if pressure and quantity of gas are held constant?

Exploring The Properties Of Gases Lab

United States. Congress. House.
Committee on Science and Technology
(2007). Subcommittee on
Investigations and Oversight

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Chemistry in the Community. American Chemical Society,2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts

ChemDiscovery Lab Manual Olga I. Agapova,2002 Exploring General, Organic, & Biochemistry in the Laboratory

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