## 8 • Electron Configuration & the Periodic Table

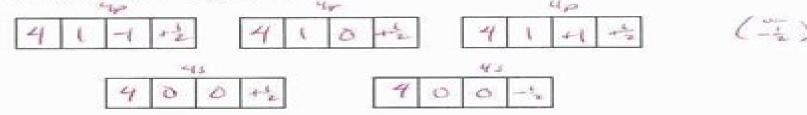
#### AP QUESTION

(a) Write the ground state electron configuration (short form or long form) for an arsenic atom.

As 152 252 206 354 306 452 3060 403

TE As [Ar] 45? 3d" 4p3 or [Ar] 3d" 452 4p3

(b) Give one permissible set of four quantum numbers for each of the outermost electrons in a single As atom when it is in its ground state.



(c) Is an isolated arsenic atom in the ground state paramagnetic or diamagnetic? Explain briefly.

Paramagnetic Holf-filled orbitale =

(d) What ion would As be expected to form? As 3.
Write the electron configuration for the arsenic ion.

As [Ar] 452 3d" 4pt or As [Ar] 3d" 452 4pt

As going electrons and has the same electron configuration as Kr.

NO F NO

P S CC 4

AS SA B. Ex

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