

For each diagram below, decide whether entropy will increase or decrease and explain your answer.

Entropy



Entropy will increase
(increase/decrease)

Entropy increases when molecules become a gas because they are more disordered and energy is spread over a larger area.



Entropy will decrease
(increase/decrease)

Entropy decreases when more order is established in a system and solids are more highly ordered than liquids.



Entropy will increase
(increase/decrease)

When temperature increases, gas molecules move even faster, colliding faster with each other, and distributing energy faster.



Entropy will increase
(increase/decrease)

When solids dissolve, the molecules spread out over a larger area and energy is distributed over a wider area, causing entropy to increase.



Entropy will increase
(increase/decrease)

Entropy tends to increase when

a) a solid is no longer a solid at the end of the reaction

In this case, the reactant Zn is a solid, but there is no



Entropy will decrease
(increase/decrease)

In this reaction, a very highly ordered solid is

produced, which has lower entropy. Also, 6 moles of liquid and 6 moles of gas combine to form a mole of

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